# organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.031 wR factor = 0.086 Data-to-parameter ratio = 16.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 1,2-Bis(2-thiazolin-2-ylsulfanyl)ethane

The title compound,  $C_8H_{12}N_2S_4$ , was synthesized by the reaction of 1,2-dibromoethane and thiazoline-2-thione. The molecule contains a centre of inversion at the mid-point of the central C-C bond.

Received 4 July 2006 Accepted 11 July 2006

## Comment

As a type of ditopic ligand, dithioethers can be used as bridging ligands in the construction of coordination polymers with soft metal ions. *N*-Heterocylic units have been synthesized and investigated (Sharma *et al.*, 1999; Constable *et al.*, 2002; Bu *et al.*, 2003; Hong *et al.*, 2000). Thiazoline derivatives possess diverse acaricide properties. Now a new thiazoline derivative, 1,2-bis(2-thiazolin-2-ylsulfanyl)ethane, (I), has been synthesized and the results are presented here.



In the molecular structure of (I), there is an inversion centre at the mid-point of the central C–C bond. Atom C3 has a distorted trigonal geometry, with the N1–C3–S2 [125.56 (17)°] and S2–C3–S1 [115.59 (12)°] angles deviating significantly from the ideal  $sp^2$ -hybridized value.

The  $Csp^2$ -S bond distances are significantly shorter than the  $Csp^3$ -S bond distances (Table 1). These values are comparable with those reported in the literature (Wang *et al.*, 2004; Wang, Zhao & Zhang, 2005; Wang, Zhao, Zheng & Duan, 2005).



## Figure 1

View of the molecule of (I) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. Atoms with suffix A are generated by the symmetry code (-x + 1, -y, -z + 1).

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## **Experimental**

A solution of 1,2-dibromoethane (1.32 g, 5 mmol) in ethanol (10 ml) was added dropwise to a mixture of thiazoline-2-thione (1.31 g, 11 mmol), KOH (0.62 g, 11 mmol) and ethanol (5 ml). The reaction mixture was then stirred for 60 h at room temperature. The precipitate was filtered off, washed with water and recrystallized from water (yield 47%, m.p. 398–399 K). Analysis calculated for  $C_8H_{12}N_2S_4$ : C 35.85, H 4.71, N 8.24%; found: C 38.89, H 4.76, N 8.16%. Crystals of (I) suitable for single-crystal X-ray analysis were grown by slow evaporation of a chloroform–ethanol (1:3  $\nu/\nu$ ) solution.

Z = 2

 $D_x = 1.484 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

Block colourless

 $0.24 \times 0.20 \times 0.16 \; \rm mm$ 

 $\mu = 0.77 \text{ mm}^-$ 

T = 293 (2) K

 $\theta_{\rm max} = 25.0^{\circ}$ 

## Crystal data

 $C_8H_{12}N_2S_4$   $M_r = 264.44$ Monoclinic,  $P2_1/c$  a = 5.8169 (13) Å b = 10.001 (2) Å c = 10.216 (2) Å  $\beta = 95.187$  (4)° V = 591.9 (2) Å<sup>3</sup>

#### Data collection

Bruker SMART CCD area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\rm min} = 0.769, T_{\rm max} = 1.000$ (expected range = 0.680–0.885)

## Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.031$   $wR(F^2) = 0.086$  S = 1.051038 reflections 64 parameters H-atom parameters constrained 2909 measured reflections 1038 independent reflections 856 reflections with  $I > 2\sigma(I)$  $R_{int} = 0.029$ 

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0429P)^{2} + 0.2261P]$ where  $P = (F_{o}^{2} + 2F_{c}^{2})/3$  $(\Delta/\sigma)_{max} < 0.001$  $\Delta\rho_{max} = 0.25 \text{ e} \text{ Å}^{-3}$  $\Delta\rho_{min} = -0.26 \text{ e} \text{ Å}^{-3}$ 

## Table 1

Selected geometric parameters (Å, °).

\$2-C3	1.753 (2)	S1-C1	1.807 (3)
S2-C4	1.819 (2)	N1-C3	1.260 (3)
\$1-C3	1.771 (2)	N1-C2	1.467 (3)
C3-S2-C4	100.68 (11)	N1-C2-C1	111.1 (2)
C3-S1-C1	89.01 (11)	N1-C3-S2	125.56 (17)
C3-N1-C2	111.51 (19)	N1-C3-S1	118.85 (17)
C2-C1-S1	105.99 (16)	S2-C3-S1	115.59 (12)

All H atoms were positioned geometrically and refined as riding (C-H = 0.97 Å), with  $U_{iso}(H) = 1.2U_{eq}(C)$ .

Data collection: *SMART* (Bruker, 1997); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

We gratefully acknowledge the 05L003 project supported by the Education Department of Liao Ning Province in China.

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